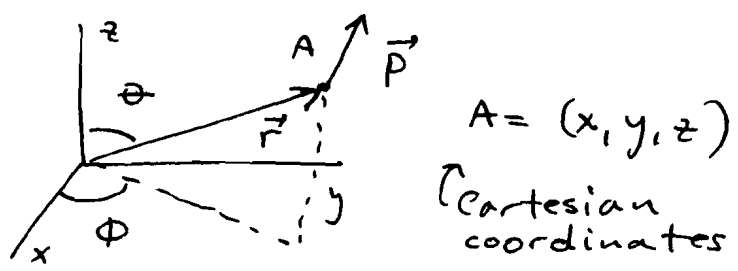


SE in 3D

$$H = \frac{\vec{p}^2}{2\mu} + V(\vec{r})$$



$A = (x, y, z)$
 ↗ Cartesian coordinates

$A = (r, \theta, \phi)$
 ↗ spherical coordinates

$$\frac{p^2}{2\mu} = -\frac{\hbar^2}{2\mu} \Delta$$

reduced mass,

$$\mu = \frac{m_1 m_2}{m_1 + m_2}$$

Example:

$$\Delta = \frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2}$$

$$\Delta(x^2 y^2 z) = 2xy^2z + 2x^2yz$$

Central potential: $V(\vec{r}) = V(r)$

example: $V(r) = -\frac{ze^2}{4\pi\epsilon_0 r}$
 ↗ Coulomb attraction

$$\Delta = \frac{\partial^2}{\partial r^2} + \frac{2}{r} \frac{\partial}{\partial r} - \frac{\vec{L}^2}{\hbar^2 r^2}$$

Angular momentum operator

$$\vec{L}^2 = -\hbar^2 \left(\frac{\partial^2}{\partial \theta^2} + \cot \theta \frac{\partial}{\partial \theta} + \frac{1}{\sin^2 \theta} \frac{\partial^2}{\partial \phi^2} \right)$$

↗ angular momentum operator transforms only spherical angles

$$-\frac{\hbar^2}{2\mu} \left(\frac{\partial^2}{\partial r^2} + \frac{2}{r} \frac{\partial}{\partial r} \right) \Psi(\vec{r}) + \frac{\vec{L}^2}{2\mu r^2} \Psi(\vec{r}) + V(r) \Psi(\vec{r}) = E \Psi(\vec{r})$$

$$\Psi(\vec{r}) = R_{n\ell}(r) \underbrace{Y_{\ell m}(\theta, \phi)}_{\text{Spherical harmonic}}$$

↑
two quantum numbers n and ℓ

$$\vec{L}^2 Y_{\ell m}(\theta, \phi) = \hbar^2 \ell(\ell+1) Y_{\ell m}(\theta, \phi)$$

$$-\frac{\hbar^2}{2\mu} \left(\frac{d^2}{dr^2} + \frac{2}{r} \frac{d}{dr} - \frac{\ell(\ell+1)}{r^2} \right) R_{n\ell}(r) + V(r) R_{n\ell}(r) = \frac{E_{n\ell}}{\text{energy spectrum}} R_{n\ell}(r)$$

For hydrogen atom:

$$E_{n\ell} = -13.6 \text{ eV} \frac{z^2}{n^2}$$

↑
energy is independent of quantum number ℓ

Quantum numbers

8-3

$$n = n_r + l + 1$$

principal quantum number \nearrow

$n_r \geq 0$ \uparrow

$l \geq 0$ \nwarrow

$l \leq n - 1$

Degeneracy: for each l there are $2l+1$ azimuthal (magnetic) quantum numbers m . They correspond to different atomic orbitals

$$Y_{l0}(\theta, \phi) \leftarrow s\text{-orbital}$$

$$Y_{1,-1}(\theta, \phi), Y_{10}(\theta, \phi), Y_{11}(\theta, \phi) \leftarrow 3\text{ p-orbitals}$$

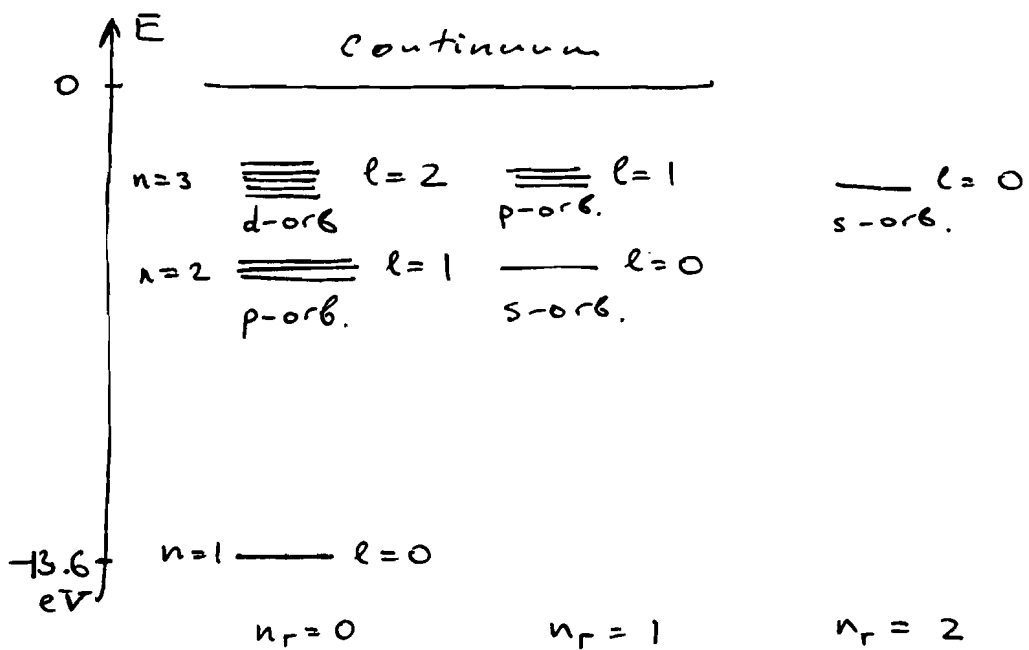
$$Y_{2,m}(\theta, \phi) \leftarrow 5\text{ d-orbitals}$$

For each given n , the overall degeneracy of l and m quantum numbers is

$$\sum_{l=0}^{n-1} (2l+1) = \underline{\underline{n^2}}$$

number of elements in a period of the periodic table

Hydrogen spectrum :



Matrix representation of operatorsHarmonic oscillator :

$$A^+ |n\rangle = \sqrt{n+1} |n+1\rangle, \quad \langle n|m\rangle = \delta_{n,m}$$

$$\langle m|A^+|n\rangle = \sqrt{n+1} \delta_{m,n+1}$$

$$\langle m|A|n\rangle = \sqrt{n} \delta_{m,n-1}$$

One may construct matrix representations for operator A :

$$A = \begin{pmatrix} 0 & \sqrt{1} & 0 & 0 & \dots \\ 0 & 0 & \sqrt{2} & 0 & \dots \\ 0 & 0 & 0 & \sqrt{3} & \dots \\ \dots & \dots & \dots & \dots & \dots \end{pmatrix}$$

Product of operators

$$\langle i|FG|j\rangle = \sum_n \langle i|F|n\rangle \langle n|G|j\rangle =$$

↑
completeness

$$= \sum_n F_{in} G_{nj}$$

↖ product of two matrices

Commutation relations follow as commutation relations of matrix product.